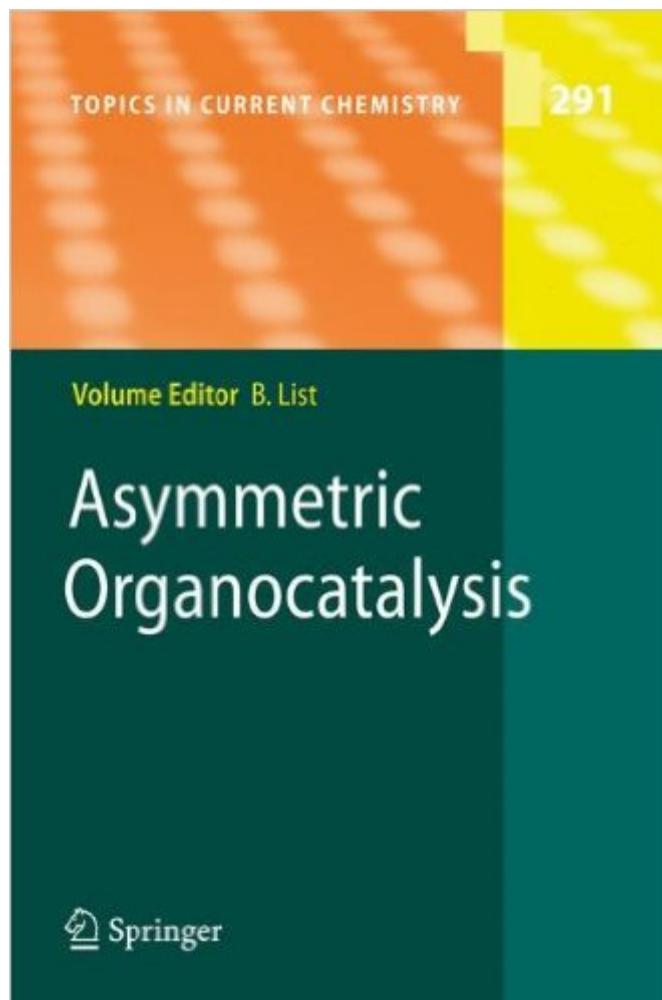


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# Asymmetric Organocatalysis (Topics In Current Chemistry)



## Synopsis

As nucleophiles, simple alkenes are typically so unreactive that only highly active electrophiles, such as carbocations, peroxides, and halogens will react with them. For the generation of carbon-carbon bonds, milder methods will often be required. Fortunately, it is possible to increase the reactivity of alkene-type *p*-nucleophiles by introducing electron-donating substituents. Substitution of one H with an OH or OR gives an enol or a vinyl ether, which are already much better nucleophiles. Using nitrogen instead of oxygen, one obtains even better nucleophiles, enamines. Enamines are among the most reactive neutral carbon nucleophiles, exhibiting rates that are even comparable to some charged nucleophiles, such as enolates [1, 2]. Most enamines, unfortunately, are sensitive to hydrolysis. The parent enamine, N, N-dimethylvinylamine, has in fact been prepared [3], but appears to be uns- ble. Enamines of cyclic ketones and many aldehydes can readily be isolated, however [4-7]. The instability of enamines might at first appear to diminish the utility of enamines as nucleophiles, but actually this property can be viewed as an added benefit: enamines can be readily and rapidly generated catalytically by using a suitable amine and a carbonyl compound. The condensation of aldehydes or ketones with amines initially affords an imine or iminium ion, which then rapidly loses a proton to afford the corresponding enamine (Scheme 1).

## Book Information

Series: Topics in Current Chemistry (Book 291)

Hardcover: 460 pages

Publisher: Springer; 2009 edition (March 4, 2010)

Language: English

ISBN-10: 3642028144

ISBN-13: 978-3642028144

Product Dimensions: 6.2 x 1.1 x 9.2 inches

Shipping Weight: 2 pounds (View shipping rates and policies)

Average Customer Review: 5.0 out of 5 starsÂ Â See all reviewsÂ (2 customer reviews)

Best Sellers Rank: #4,846,260 in Books (See Top 100 in Books) #112 inÂ Books > Science & Math > Chemistry > Organic > Organometallic Compounds #1156 inÂ Books > Science & Math > Chemistry > Inorganic #2831 inÂ Books > Science & Math > Chemistry > Industrial & Technical

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Great book for the practitioner of organocatalysis. This book really puts the field in perspective. The

best part about the book is the classification of reaction types which makes it tremendously useful for efforts at new reaction discovery.

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